

A CONVENIENT NEW ROUTE FROM DIPHOSPHENE TO 1,3-DIPHOSPHA-ALLENE AND DYNAMIC
NMR STUDIES OF THE 2,4,6-TRI-*t*-BUTYLPHENYL DERIVATIVE

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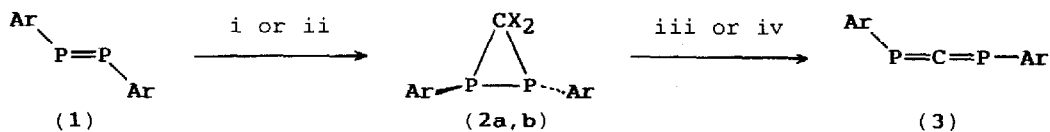
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Abstract: Addition of dihalocarbenes to a diphosphene gave diphosphiranes which undergo rearrangement on treatment with alkyllithium reagents to afford 1,3-diphospha-allene; dynamic NMR studies on the allene are described.

1,3-Diphospha-allene is a new type of unusual organophosphorus compound of a cumulene frame having two phosphorus atoms in low coordination state protected by steric bulkiness. We have reported the first preparation and characterization of 1,3-bis(2,4,6-tri-*t*-butylphenyl)-1,3-diphospha-allene (3)¹⁾ via silyloxyphosphinophospha-ethylene.²⁾ Karsch³⁾ and Appel⁴⁾ also reported the other preparative methods for 3.

We now report a new convenient method for 3 from the E-diphosphene 1.⁵⁾ Our present method involves addition of dihalocarbenes⁶⁾ to the P=P double bond of the diphosphene 1, followed by ring opening of the resulting diphosphiranes 2 with alkyllithium reagents. The latter ring opening reactions have often been used for the preparation of allenes.^{7,8)} The diphosphirane 2a (0.0761 mmol) in 4.5 ml of ether (δ_p -69.5) was allowed to react with methyllithium (0.107 mmol) in ether (0.1 ml) at -78 °C and was immediately warmed up to room temperature with stirring for 45 min to give 62% yield of pure 3 (δ_p 141.7) after column chromatography (silica gel, pentane). The whole process could be carried out in a "one-pot" manner without isolation of 2. For example, in the case of dibromodiphosphirane, 2b (δ_p -70.3) was not purified for the further reaction using THF as a solvent and butyllithium in hexane was used for the final step to give 20% yield of 3 with 76% recovery of the diphosphene 1 after chromatographic separation. All spectral data including ¹H, ¹³C, ³¹P NMR and MS of 3 thus obtained were identical with those of an authentic sample.¹⁾

We have observed racemization of 3 around the P=C=P axis⁹⁾ with a mercury



Ar = 2,4,6-Bu^t₃C₆H₂; a: X=Cl, b: X=Br. Reagents: i, CHCl₃ + KOBu^t or BuⁿLi;
ii, CHBr₃ + KOBu^t or CBr₄ + BuⁿLi; iii, MeLi; iv, BuⁿLi.

lamp and intramolecular cyclization reaction²⁾ on heating in boiling toluene. We were successful in the first observation on the internal rotation about the P-C_{Ar} bond of this particular compound 3. ¹H NMR spectra of 3 in CDCl₃ were obtained on a JEOL GX-270 at 270.166 MHz between at -60.0 and 56.7 °C. At -60.0 °C, o-t-butyl groups appeared as two singlets at δ 1.60 and 1.10 together with that of p-t-butyl at δ 1.29 in almost equal intensities. At 24 °C the o-t-butyl signals started to coalesce and finally at 56.7 °C they became a broad singlet at δ 1.36 whereas the signal due to p-t-butyl remained constant in shape as a sharp singlet at δ 1.29 during the temperature range mentioned above. By the method of line shape analysis,¹⁰⁾ the rate constants (k) were obtained for the internal rotation about the P-C_{Ar} bond.¹¹⁾ From the plot of ln(k/T) vs. 1/T, at 8 different temperatures between 0.0 and 56.7 °C, the activation parameters are as follows (correlation factor for the Eyring plot r = 0.9998): ΔH[‡]₂₇₃ = 12.3 ± 0.2 kcal/mol and ΔS[‡]₂₇₃ = -6.4 ± 0.8 e.u. The first order rate constant k is 30.4 s⁻¹ at 0 °C. Taking the Dreiding stereomodel of 3 into account, the two adjacent bulky aryl groups with o-t-butyl cogs might correlate each other to rotate in a gear mode.¹²⁾

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